

Tissue Iron Content Assay Kit

Note: Take two or three different samples for prediction before test.

Operation Equipment: Spectrophotometer

Cat Number: NA0302

Size: 50T/48S

Components:

Extract solution: Liquid 55 mL×1. Storage at 4°C.

Reagent I: Powder×2. Storage at 4°C. Add 7.5 mL of distilled water before use. [Prepare the reagent when it will be used. When the reagent turns black, it cannot be used;](#)

Reagent II: Powder×2. Storage at 4°C. Add 375 µL of glacial acetic acid and 12 mL of distilled water before use. Unused reagent can be stored for one week at 4 °C.

Standard Solution: Liquid 3 mL×1, 1 µmol/mL Fe³⁺ standard solution. Storage at 4°C. Add distilled water dilute 8 times to form a standard solution of 0.125 µmol/mL before use. [Prepare when the solution will be used.](#)

Product Description:

Iron is one of the essential trace elements in human body, which is the main component of hemoglobin, myoglobin, cytochrome and other enzyme systems. Iron can assist in the transport of oxygen and promote fat oxidation. Iron deficiency can easily cause anemia, metabolic disorders, and affect the immune function of the body.

Fe³⁺ is reduced by sodium sulfite to Fe²⁺, which reacts with 2,2-dipyridine-bipyridine, have an absorption peak at 520 nm. According measure absorbance at 520 nm can reflect tissue iron concentration.

Reagents and Equipment Required but Not Provided.

Spectrophotometer, centrifuge, chloroform, adjusted transferpettor, [mortar/homogenizer](#), 1 mL glass cuvette, chloroform, ice and distilled water.

Procedure:

Add 1 mL of Extract solution to 0.1 g of tissue, and fully homogenize on ice bath. Centrifuge at 4000×g for 10 minutes at 4°C to remove insoluble materials, and take the supernatant for test.

Detection:

1. Preheat the spectrophotometer for 30 minutes, adjust wavelength to 520 nm, set zero with distilled water.
2. Add reagents with the following list:

Reagent Name (µL)	Blank tube (A _B)	Test tube (A _T)	Standard tube (A _S)
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Distilled water	400	-	-
Standard solution (0.125 $\mu\text{mol/mL}$)	-	-	400
Sample	-	400	-
Reagent I	200	200	200
Reagent II	400	400	400

Mix thoroughly, incubate in boiling water bath for 5 minutes, cooling liquid. Add 200 μL of chloroform. Shake well and centrifuge at 10000 rpm for 10 minutes at room temperature. Take 700 μL of supernatant to 1 mL glass cuvette. Measure absorbance at 520 nm. Recorded as A_B , A_T , A_S . Calculate $\Delta A_T = A_T - A_B$, $\Delta A_S = A_S - A_B$.

III. Calculation

1) Tissue weight

$$\text{Tissue iron } (\mu\text{g/g}) = C_s \times \Delta A_T \div \Delta A_S \times V_e \times 55.845 \div W = 6.98 \times \Delta A_T \div \Delta A_S \div W$$

2) Tissue protein concentration

$$\text{Tissue iron } (\mu\text{g/mg prot}) = C_s \times \Delta A_T \div \Delta A_S \times V_e \times 55.845 \div (C_{pr} \times V_e) = 6.98 \times \Delta A_T \div \Delta A_S \div C_{pr}$$

C_s : Fe^{3+} standard solution, 0.125 $\mu\text{mol/mL}$;

55.845: Relative molecular mass of Fe, 55.845 $\mu\text{g}/\mu\text{mol}$;

V_e : Extract solution volume, 1 mL;

C_{pr} : Supernatant sample protein concentration (mg/mL);

W : Sample weight, g.

Note:

1. When $\Delta A > 1$, please dilute the sample to appropriate concentration with distilled water, multiply dilute times in the formula.
2. Reagent I cannot be used if it becomes black after dissolution. Reagent II is toxic, take self-protection measures when using.

Related products :

NA0661/NA0420 Serum Total Iron Binding Capacity(TIBC) Assay Kit

NA0664/NA0422 Water Chromium(VI) Content Assay Kit

NA0663/NA0421 Phosphate Content Assay Kit

NA0662/NA0379 Total Phosphorus Content Assay Kit

Technical Specifications :

Minimum Detection Limit: 0.00009449 $\mu\text{mol/mL}$

Linear Range: 0.0039-0.25 $\mu\text{mol/mL}$