

# Hydrogen sulfide (H<sub>2</sub>S) Content Assay Kit

**Note:** Take two or three different samples for prediction before test.

**Operation Equipment:** Spectrophotometer/Microplate reader

**Catalog Number:** NA0480

**Size:** 100T/96S

## Components:

Reagent	Size	Storage
Extract solution I	Solution 110 mL×1	4°C
Extract solution II	Solution 20 mL×1	4°C
Reagent I	Solution 8 mL×1	4°C
Reagent II	Solution 8 mL×1	4°C

## Product Description:

Hydrogen sulfide (H<sub>2</sub>S) is a new type of gaseous signal molecule. It is a neurotransmitter that exists in the brain. The physiological concentration of H<sub>2</sub>S has an important regulatory effect on the long-term enhancement of the hippocampus of the nervous system. It also plays an important pathophysiological effect on the process of spontaneous hypertension, hemorrhagic shock and liver cirrhosis.

H<sub>2</sub>S can react with N, N-dimethyl-p-phenylenediamine and ferric ammonium sulfate to form methylene blue. Methylene blue has a maximum absorption peak at 665nm. The H<sub>2</sub>S content can be calculated by measuring the absorbance value.

## Reagents and Equipment Required but Not Provided:

Spectrophotometer/microplate reader, desk centrifuge, pipette, micro glass cuvette/96 well flat-bottom plate, mortar/homogenizer, ice and distilled water.

## Procedure

### I. Sample preparation:

#### 1. Bacteria or cells

Bacteria or cells: collecting bacteria or cells into the centrifuge tube, centrifugation and discard supernatant. Suggest add 1 mL of Extract solution I to 5 million of bacteria or cells. Use ultrasonication to splitting bacteria and cell (placed on ice, ultrasonic power 200W, ultrasonic 3 seconds, interval 7 seconds, total time 3 minutes). Centrifuge at 12000 ×g for 10 minutes at 4°C to remove insoluble materials. Then add 0.15 mL Extract solution II to 0.8 mL supernatant. Centrifuge at 12000 ×g for 10 minutes at 4°C to remove insoluble materials. Take the supernatant on ice for test.

2. Tissue: add 1 mL of Extract solution I into 0.1 g of tissue and fully grind on ice. Centrifuge at 12000 ×g for 10minutes at 4°C to remove insoluble materials. Then add 0.15 mL Extract solution II to 0.8 mL supernatant. Centrifuge at 12000 ×g for 10 minutes at 4°C to remove insoluble materials. Take the supernatant on ice for test.

3. Serum (plasma) or other liquid samples: add 1 mL of Extract solution I into 0.1 mL of serum (plasma). Centrifuge at 12000 ×g for 10 minutes at 4°C to remove insoluble materials. Then add 0.15 mL Extract solution II to 0.8 mL supernatant. Centrifuge at 12000 ×g for 10 minutes at 4°C to remove insoluble materials. Take the supernatant on ice for test.

## II. Determination procedure:

1. Preheat spectrophotometer/microplate reader for 30min, adjust wavelength to 665 nm, set zero with distilled water.
2. Determination:

Reagent (μL)	Test tube	Blank tube
Sample	50	
Distilled water		50
Reagent I	75	75
Reagent II	75	75
Mix well. React at room temperature for 10 minutes. Measure the absorbance at 665 nm, record as A <sub>T</sub> , A <sub>B</sub> . Calculate the ΔA=A <sub>T</sub> -A <sub>B</sub> .		

Note: blank tube only need to be test one or two times.

## III. Calculations:

A. 96 well flat-bottom plate

Take the concentration of standard solution(nmol/mL) as x-axis, and the corresponding ΔA is y-axis. Then the linear regression equation  $y = 0.0020x - 0.0633$  is obtained. Bring ΔA into the equation to get x (nmol/mL).

1. Protein concentration:

$$\text{H}_2\text{S content (nmol/mg prot)} = x \times V_S \div (V_S \times C_{pr}) = x \div C_{pr}$$

2. Sample weight:

$$\text{H}_2\text{S content (nmol/g weight)} = x \times (V_{SP} + V_{EX2}) \div (W \times V_{SP} \div V_{EX1}) = 1.1875 \times x \div W$$

3. Cell amount:

$$\text{H}_2\text{S content (nmol/10}^4 \text{ cell)} = x \times (V_{SP} + V_{EX2}) \div (\text{cells} \times V_{SP} \div V_{EX1}) = 1.1875 \times x \div \text{cells}$$

4. Serum (plasma) sample:

$$\text{H}_2\text{S content (nmol/mL)} = x \times (V_{SP} + V_{EX2}) \div [(V_L \times V_{SP} \div (V_{EX1} + V_L))] = 13.0625 \times x$$

V<sub>s</sub>: Sample volume in reaction, 0.05 mL;

V<sub>SP</sub>: Supernatant volume in Extraction, 0.8 mL;

V<sub>EX1</sub>: Extraction solution I volume, 1 mL;

V<sub>EX2</sub>: Extraction solution II volume, 0.15 mL;

C<sub>pr</sub>: Sample protein concentration, mg/mL;

W: Sample weight, g;

cells: Total number of bacteria and cells, 10<sup>4</sup>;

V<sub>L</sub>: Liquid sample volume, 0.1 mL.

## B. Micro glass cuvette

Take the concentration of standard solution (nmol/mL) as x-axis, and the corresponding  $\Delta A$  is y-axis. Then the linear regression equation  $y = 0.0026x - 0.0268$  is obtained. Bring  $\Delta A$  into the equation to get x (nmol/mL).

### 1. Protein concentration:

$$\text{H}_2\text{S content (nmol/mg prot)} = x \times V_S \div (V_S \times C_{pr}) = x \div C_{pr}$$

### 2. Sample weight:

$$\text{H}_2\text{S content (nmol/g weight)} = x \times (V_{SP} + V_{EX2}) \div (W \times V_{SP} \div V_{EX1}) = 1.1875 \times x \div W$$

### 3. Cell amount:

$$\text{H}_2\text{S content (nmol}/10^4 \text{ cell)} = x \times (V_{SP} + V_{EX2}) \div (\text{cells} \times V_{SP} \div V_{EX1}) = 1.1875 \times x \div \text{cells}$$

### 4. Serum (plasma) sample:

$$\text{H}_2\text{S content (nmol/mL)} = x \times (V_{SP} + V_{EX2}) \div [(V_L \times V_{SP} \div (V_{EX1} + V_L))] = 13.0625 \times x$$

$V_S$ : Sample volume in reaction, 0.05 mL;

$V_{SP}$ : Supernatant volume in Extraction, 0.8 mL;

$V_{EX1}$ : Extraction solution I volume, 1 mL;

$V_{EX2}$ : Extraction solution II volume, 0.15 mL;

$C_{pr}$ : Sample protein concentration, mg/mL;

$W$ : Sample weight, g;

cells: Total number of bacteria and cells,  $10^4$ ;

$V_L$ : Liquid sample volume, 0.1 mL.

## Note:

1. If the  $\Delta A$  is lower, it is recommended to increase the sample size before determination; If  $\Delta A > 1.2$ , it is recommended to dilute the sample before determination. The calculation formula should be multiplied by the corresponding dilution factor.

## Examples:

1. Take 0.1g of mouse liver to follow the determination procedure to operate. Determination with 96 well flat-bottom plate, and calculate  $\Delta A = A_T - A_B = 0.150 - 0.121 = 0.029$ . The calculated content is as follows:

$$\text{H}_2\text{S content (nmol/g weight)} = x \times (V_{SP} + V_{EX2}) \div (W \times V_{SP} \div V_{EX1}) = 548.03 \text{ nmol/g weight.}$$

2. Take 0.1g of purple leaf plum to follow the determination procedure to operate. Determination with 96 well flat-bottom plate, and calculate  $\Delta A = A_T - A_B = 0.270 - 0.121 = 0.149$ . The calculated content is as follows:

$$\text{H}_2\text{S content (nmol/g weight)} = x \times (V_{SP} + V_{EX2}) \div (W \times V_{SP} \div V_{EX1}) = 1260.53 \text{ nmol/g weight.}$$

## Related products:

NA0781/ NA0540 Reduced Glutathione (GSH) Assay Kit

NA0780/ NA0539 Oxidized Glutathione (GSSG) Assay Kit

NA0779/ NA0538    Glutathione Peroxidase (GPX) Assay Kit  
NA0783/ NA0542    Oxidized Thioredoxin Reductase (TrxR) Assay Kit  
NA0752/ NA0510    Nitric oxide (NO) Assay Kit