

# Water Mercury Ion (Hg<sup>2+</sup>) Content Assay Kit

**Note:** Take two or three different samples for prediction before test.

**Operation Equipment:** Spectrophotometer

**Cat Number:** NA0665

**Size:** 50T/48S

## Components:

Reagent I: Powder ×1. Storage at 4°C. Dissolve with 2 mL of distilled water before use.

Reagent II: Liquid 5 mL×1. Storage at 4°C.

Reagent III: Liquid 12 mL×1. Storage at 4°C.

Reagent IV: Powder ×1. Storage at 4°C. Dissolve with 5 mL of distilled water before use.

Reagent V: Powder ×1. Storage at 4°C and protect from light. Dissolve with 50 mL of chloroform (**self-provided**) before use.

Reagent VI: Liquid 20 mL×1. Storage at 4°C.

Standard Solution: Liquid 1 mL×1, 4000 nmol/mL Hg<sup>2+</sup> standard solution. Storage at room temperature. Add distilled water dilute 400 times to form a standard solution of 10 nmol/mL before use.

## Product Description:

Hg<sup>2+</sup> is an important toxic heavy metal ion in water, which easily absorbed and accumulated by organisms and can be further transmitted through the food chain, causing damage. Minamata disease is a kind of typical mercury poisoning.

After digestion, Hg<sup>2+</sup> can form one orange complex with Dithizone in acid environment, which can be dissolved in chloroform and has a maximum absorption peak at 490 nm. In this kit, the content of Hg<sup>2+</sup> is quantified by measuring the color development at 490 nm.

## Reagents and Equipment Required but Not Provided.

Spectrophotometer, centrifuge, adjusted transferpettor, 1 mL glass cuvette, concentrated sulfuric acid (H<sub>2</sub>SO<sub>4</sub>), concentrated nitric acid (HNO<sub>3</sub>), distilled water.

## Procedure:

Add 7 mL of concentrated nitric acid immediately after every 1000 mL of water sample is collected. Adjust the pH ≤ 1. If the water sample cannot be measured immediately after sampling, add 4 mL or more of Reagent II to each liter of sample to make it lasting pale red.

## Detection:

1. Preheat the spectrophotometer for 30 minutes, adjust wavelength to 490 nm, set zero with chloroform.
2. Add reagents with the following list to 5 mL EP tubes:

| Reagent Name ( $\mu\text{L}$ )   | Test tube ( $A_T$ ) | Standard tube ( $A_S$ ) | Blank tube ( $A_B$ ) |
|--|---------------------|-------------------------|----------------------|
| Water sample   | 1000                | -                       | -                    |
| Standard solution  | -                   | 1000                    | -                    |
| Distilled water  | -                   | -                       | 1000                 |
| Concentrated sulfuric acid   | 40                  | 40                      | 40                   |
| Concentrated nitric acid   | 10                  | 10                      | 10                   |
| Reagent I  | 32                  | 32                      | 32                   |
| Reagent II   | 60                  | 60                      | 60                   |
| Seal with parafilm, mix thoroughly and shock 2 minutes. Digest in $95^\circ\text{C}$ water bath for 2 hours, then cool to about $40^\circ\text{C}$ .   |                     |                         |                      |
| Reagent III  | 200                 | 200                     | 200                  |
| Shake until the solution in the EP tube is clear and transparent. Open the lid and leave for 10 minutes. Shake several times during standing to allow the gas escape.  |                     |                         |                      |
| Reagent IV   | 80                  | 80                      | 80                   |
| Reagent V  | 1000                | 1000                    | 1000                 |
| Fully shake for 2 minutes after capping, let stand for 10 minutes. Suck the organic phase in the lower layer into 1.5mL EP tubes.  |                     |                         |                      |
| Reagent VI   | 400                 | 400                     | 400                  |
| Fully shake to make the organic phase without green. After standing and delaminating, absorb the organic phase and measure the absorbance at 490 nm. Recorded as $A_T$ , $A_S$ , $A_B$ . $\Delta A_T = A_T - A_B$ , $\Delta A_S = A_S - A_B$ . |                     |                         |                      |

### Calculations

$$\text{Hg}^{2+} (\text{nmol/mL}) = C_s \times \Delta A_T \div \Delta A_S = 10 \times \Delta A_T \div \Delta A_S$$

$C_s$ :  $\text{Hg}^{2+}$  standard solution (10 nmol/mL).

### Note:

- 1000  $\mu\text{g/L}$  copper ion, 20  $\mu\text{g/L}$  silver ion, 10  $\mu\text{g/L}$  gold ion, 5  $\mu\text{g/L}$  platinum ion in water sample without interference.
- Pay attention to safety during measurement, wear masks and gloves to avoid inhalation or exposure to toxic and dangerous reagents.
- When the absorbance is greater than 1, please dilute the serum to appropriate concentration with distilled water.
- Water with less suspended matter and/or organic matter can shorten the heating time to 1 hour, and clean water without suspended matter can shorten the heating time to 30 minutes.
- if the upper solution of sample tube becomes transparent during digestion, Reagent II can be added appropriately to keep the sample tube pink or black purple.
- if the added Reagent III is not enough to make the sample tube clear, the amount of Reagent III can be added to make the sample tube clear.

7. if the lower organic phase still appears green after Reagent VI is added, the amount of Reagent VI can be increased to make the organic phase transition of the lower layer shallow.

**Related Products:**

NA0664/NA0422 Water Chromium( $\text{Cr}^{6+}$ ) Content Assay Kit  
NA0662/NA0379 Total Phosphorus Content Assay Kit  
NA0302/NA0301 Tissue Iron Content Assay Kit  
NA0296/NA0295 Blood Ammonia Content Assay Kit